

Europium antimony sulfide, $\text{Eu}_6\text{Sb}_6\text{S}_{17}$ Gengbang Jin,^a Daniel M. Wells,^a
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Key indicators

Single-crystal X-ray study
 $T = 193 \text{ K}$
Mean $\sigma(\text{S}-\text{S}) = 0.003 \text{ \AA}$
 R factor = 0.031
 wR factor = 0.062
Data-to-parameter ratio = 26.8For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

$\text{Eu}_6\text{Sb}_6\text{S}_{17}$ is isotypic with $\text{Sr}_6\text{Sb}_6\text{S}_{17}$ and can be regarded as consisting of two isolated $[\text{Sb}_3\text{S}_7]^{5-}$ anions and an S_3^{2-} polysulfide anion that are joined together by Eu^{2+} cations. There are six crystallographically unique Eu^{2+} cations bound by sulfide and polysulfide anions in seven-, eight- and nine-coordinate environments. Each of the $[\text{Sb}_3\text{S}_7]^{5-}$ anions consists of a trinuclear assembly of corner-sharing SbS_3 units. The $[\text{Sb}_3\text{S}_7]^{5-}$ anions are connected by long (3.1 Å) $\text{Sb}-\text{S}$ interactions, forming one-dimensional ribbons running down the a axis and packed with opposing directions of polarity. The bent S_3^{2-} anions stack in columns parallel to the a axis, oriented in opposing directions with respect to one another. The overall structure is three-dimensional and has channels running down the a axis to house the stereochemically active lone pair of electrons on the Sb^{3+} centers. The presence of Eu^{2+} is supported by both magnetic measurements and bond-valence calculations.

Comment

Ternary rare-earth thioantimonites have been known for more than 20 years, since the initial report of the synthesis and crystal structure of $\text{Eu}_3\text{Sb}_4\text{S}_9$ (Lemoine *et al.*, 1981). Despite this early success, the chemistry of thioantimonites is relatively undeveloped compared to that of thiophosphates (Evenson & Dorhout, 2001; Kanatzidis, 1997), and there is still a dearth of understanding in the structure–property relationships in this system. There has been renewed interest in ternary and quaternary chalcantimonites and chalcobismuthites owing to their potential applications as thermoelectric materials. Based on structural analogies with Pn_2Q_3 phases ($\text{Pn} = \text{Sb}$ and Bi , and $\text{Q} = \text{S}$, Se and Te), a series of ternary phases such as EuSb_2Q_4 , EuSb_2Q_7 , and EuBi_2Te_4 has been prepared and their transport properties measured (Godzhaev *et al.*, 1978; Harman & Honig, 1967). EuBi_2Te_4 displays a thermoelectric figure of merit comparable to that of Bi_2Te_3 (Godzhaev *et al.*, 1978; Harman & Honig, 1967). $\text{Eu}_2\text{Bi}_4\text{S}_4$ is also of particular interest because it contains Eu^{II} and Eu^{III} in different coordination environments (Carré *et al.*, 1985). Other ternary rare-earth thioantimonites, such as $\text{Pr}_8\text{Sb}_2\text{S}_{15}$ (Guseinov *et al.*, 1981), $\text{Ln}_3\text{Sb}_3\text{S}_{10}$ ($\text{Ln} = \text{La}$ and Ce ; Gao *et al.*, 1983) and $\text{Ln}_6\text{Sb}_8\text{S}_{21}$ ($\text{Ln} = \text{La}$ and Ce ; Gao *et al.*, 1984), have been characterized from single-crystal and powder X-ray diffraction data. EuSb_4S_7 has also been reported, but only its lattice constants are known (Aliev *et al.*, 1978).

More recently, there has been attention paid to preparing quaternary rare-earth chalcantimonites and chalcobismuthites, *e.g.* $\text{K}_2(\text{Ln}_{2-x})\text{Sb}_{4+x}\text{Sb}_4\text{Se}_{12}$ ($\text{Ln} = \text{La}$, Ce , Pr and Gd ; Chen & Dorhout, 1997), $\text{K}_2\text{Gd}_2\text{Sb}_2\text{Se}_9$ (Choi *et al.*, 1999), $\text{K}_2\text{La}_2\text{Sb}_2\text{S}_9$ (Choi *et al.*, 1999), $\text{Na}_9\text{Gd}_5\text{Sb}_8\text{S}_{26}$ (Park & Kim,

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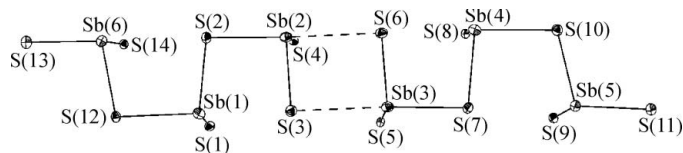


Figure 1

A view of the $[\text{Sb}_3\text{S}_7]^{5-}$ anions, which consist of trinuclear assemblies of corner-sharing SbS_3 units in $\text{Eu}_6\text{Sb}_6\text{S}_{17}$. 50% probability displacement ellipsoids are depicted.

2001), $\text{BaLaBi}_2\text{S}_6$ (Choi *et al.*, 1997), $\text{ALn}_{1\pm x}\text{Bi}_{4\pm x}\text{S}_8$ ($A = \text{K}$ and Rb , and $\text{Ln} = \text{La}$, Ce , Pr and Nd ; Iordanidis *et al.*, 1999) and $\text{Pb}_2\text{Eu}_2\text{Bi}_6\text{Se}_{13}$ (Chung *et al.*, 1998). These compounds possess complex structures, owing to the combination of the high coordination numbers found for lanthanides with discrete $[\text{PnQ}_3]^{3-}$, $[\text{PnQ}_{3+1}]^{3-}$ and $[\text{PnQ}_4]^{5-}$ anions and the extended networks of chalcantimonites and chalcobismuthites. We report here the preparation and crystal structure of $\text{Eu}_6\text{Sb}_6\text{S}_{17}$, a new ternary thioantimonite that is isotypic with $\text{Sr}_6\text{Sb}_6\text{S}_{17}$ (Choi & Kanatzidis, 2000) and homeotypic with $\text{Pb}_6\text{Sb}_6\text{S}_{17}$ (with one of the axes halved) (Orlandi *et al.*, 2002), and which contains Eu^{II} and a polysulfide linkage.

Although three-dimensional in nature, the structure of $\text{Eu}_6\text{Sb}_6\text{S}_{17}$ can be treated as consisting of two isolated $[\text{Sb}_3\text{S}_7]^{5-}$ anions and an S_3^{2-} polysulfide anion that are joined together by Eu^{2+} cations. Analogous to $\text{Sr}_6\text{Sb}_6\text{S}_{17}$ (Choi & Kanatzidis, 2000), charge balance is attained in a straightforward manner by assuming the presence of six Eu^{2+} cations, six Sb^{3+} cations, 14 S^{2-} anions and one S_3^{2-} anion. The $[\text{Sb}_3\text{S}_7]^{5-}$ anion consists of a trinuclear assembly of corner-sharing SbS_3 units, as shown in Fig. 1. Other anions formed from the linking of SbS_n units include $[\text{Sb}_2\text{S}_4]^{2-}$ in BaSb_2S_4 (Cordier *et al.*, 1984) and $[\text{Sb}_2\text{S}_5]^{4-}$ in $\text{Sr}_2\text{Sb}_2\text{S}_5 \cdot 15\text{H}_2\text{O}$ (Schwidetzky, 1985). The $\text{Sb}-\text{S}$ bond distances range from 2.416 (2) to 2.664 (3) Å. As in $\text{Sr}_6\text{Sb}_6\text{S}_{17}$ (Choi & Kanatzidis, 2000), the $[\text{Sb}_3\text{S}_7]^{5-}$ anions are connected by long 3.105 (3) to 3.106 (3) Å $\text{Sb} \cdots \text{S}$ interactions to form one-dimensional ribbons running down the a axis. Each of the Sb^{3+} cations possesses a stereochemically active lone pair of electrons.

Although the space group $P2_12_12_1$ in which this compound crystallizes is non-centrosymmetric but non-polar, there is approximate alignment of the lone pair of electrons on the individual Sb^{3+} cations parallel to the b axis. However, the $[\text{Sb}_3\text{S}_7]^{5-}$ anions are packed with opposing directions of polarity (*i.e.* the lone-pairs on different anions are aligned in opposite directions). The S_3^{2-} anion is bent, with $\text{S}-\text{S}$ bond distances of 2.091 (3) and 2.092 (3) Å, and an $\text{S}-\text{S}-\text{S}$ bond angle of 113.34 (15)°. Although the polysulfide anions do stack in oriented columns parallel to the a axis, the columns are oriented in opposing directions with respect to one another, canceling any polarity.

The six crystallographically unique Eu^{2+} cations are found in a variety of coordination environments, as shown in Fig. 2. Atoms $\text{Eu}1$ and $\text{Eu}2$ are found in seven-coordinate monocapped trigonal prismatic environments, with $\text{Eu}-\text{S}$ bond distances ranging from 2.984 (3) to 3.081 (3) Å, and 2.944 (3) to 3.209 (3) Å, respectively. Atom $\text{Eu}3$ is bound by eight

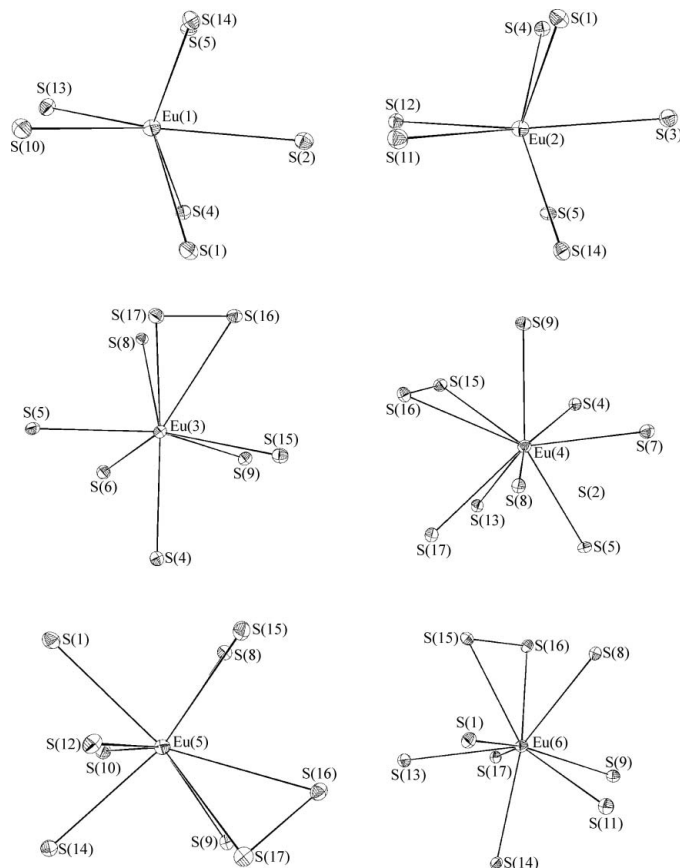


Figure 2

A depiction of the local environments of the six crystallographically unique Eu^{2+} cations in $\text{Eu}_6\text{Sb}_6\text{S}_{17}$. 50% probability displacement ellipsoids are depicted.

sulfide anions, with bond distances ranging from 2.968 (2) to 3.235 (3) Å. The geometry around atom $\text{Eu}3$ is heavily distorted from being a bicapped trigonal prism, a square antiprism or a dodecahedron because two of the sulfide nearest neighbors are part of the polysulfide linkage, creating a very short edge on the polyhedron. Atoms $\text{Eu}4$, $\text{Eu}5$ and $\text{Eu}6$ are all in nine-coordinate geometries that are quite distorted from idealized tricapped trigonal prismatic symmetry because of the presence of a polysulfide linkage within their inner spheres. The $\text{Eu}4-\text{S}$, $\text{Eu}5-\text{S}$ and $\text{Eu}6-\text{S}$ bond distances range from 3.005 (3) to 3.382 (3) Å, 2.980 (2) to 3.208 (2) Å, and 2.988 (3) to 3.197 (3) Å, respectively. Bond-valence sum calculations provide values ranging from 1.87 to 2.10 for the Eu centers in $\text{Eu}_6\text{Sb}_6\text{S}_{17}$, which are consistent with Eu^{2+} in this compound (Brown & Altermatt, 1985; Brese & O'Keeffe, 1991). Measurements of the magnetic susceptibility show that the effective magnetic moment is 7.0 μ_B/Eu , which is somewhat lower than the expected free-ion moment of 7.9 μ_B for Eu^{2+} because of the presence of minor amounts of Sb_2S_3 impurities. An antiferromagnetic transition occurs at 3.2 K in the susceptibility curve. A packing diagram for the complete structure of $\text{Eu}_6\text{Sb}_6\text{S}_{17}$ is shown in Fig. 3. When the structure is viewed down the a axis, it becomes apparent that $\text{Eu}_6\text{Sb}_6\text{S}_{17}$ adopts a channel structure. These

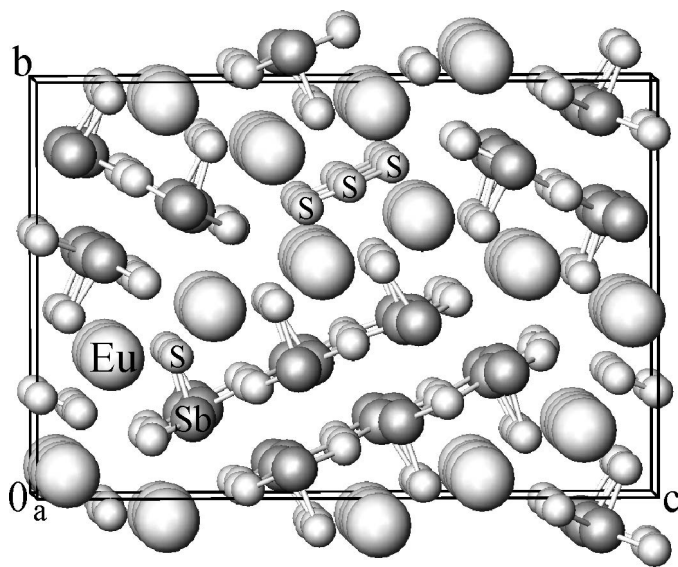


Figure 3
The three-dimensional channel structure of $\text{Eu}_6\text{Sb}_6\text{S}_{17}$ viewed down the a axis. Eu–S bonds have been omitted for clarity.

channels house the stereochemically active lone pair of electrons on the Sb^{3+} centers. A similar channel motif is also observed for $\text{Eu}_3\text{Sb}_4\text{S}_9$ (Lemoine *et al.*, 1981).

Experimental

Eu foil (99.9%, Alfa-Aesar), Sb (99.5%, Alfa-Aesar) and S (99.5%, Alfa-Aesar) were loaded in a fused-silica tube in a molar ratio of 3:5:12 with a total mass of 0.250 g. The following heating profile was used: 5 K min^{-1} to 1123 K for 1 d, 5 K min^{-1} to 1273 K for 7 d, 0.5 K min^{-1} to 873 K for 5 d, and 0.25 K min^{-1} to 295 K. The product consisted of small black block-shaped crystals of $\text{Eu}_6\text{Sb}_6\text{S}_{17}$ and acicular crystals of Sb_2S_3 . The yields were generally low (about 10%), and were not improved by quenching the reactions. Semi-quantitative SEM/EDX analysis was performed on crystals of $\text{Eu}_6\text{Sb}_6\text{S}_{17}$ using a Jeol 840/Link Isis instrument. Eu, Sb and S percentages were calibrated against standards, and a Eu:Sb:S ratio of close to 1:1:2.8 (20 mol%:23 mol%:57 mol%) was found.

Crystal data

$\text{Eu}_6\text{Sb}_6\text{S}_{17}$
 $M_r = 2187.28$
Orthorhombic, $P2_12_12_1$
 $a = 8.236$ (2) Å
 $b = 15.237$ (3) Å
 $c = 22.724$ (5) Å
 $V = 2851.7$ (11) Å³
 $Z = 4$
 $D_x = 5.095$ Mg m^{-3}

Mo $K\alpha$ radiation
Cell parameters from 7062 reflections
 $\theta = 1.6$ – 28.3°
 $\mu = 19.80$ mm^{-1}
 $T = 193$ (2) K
Prism, black
 $0.11 \times 0.08 \times 0.05$ mm

Data collection

Bruker SMART APEX diffractometer
 ω scans
Absorption correction: numerical (SHELXTL; Sheldrick, 1997)
 $T_{\text{min}} = 0.102$, $T_{\text{max}} = 0.383$
29271 measured reflections

7062 independent reflections
6058 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.055$
 $\theta_{\text{max}} = 28.3^\circ$
 $h = -10 \rightarrow 10$
 $k = -20 \rightarrow 19$
 $l = -30 \rightarrow 30$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.031$
 $wR(F^2) = 0.062$
 $S = 1.00$
7062 reflections
264 parameters
 $w = 1/[\sigma^2(F_o^2) + (0.0189P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.002$
 $\Delta\rho_{\text{max}} = 1.41$ e Å⁻³
 $\Delta\rho_{\text{min}} = -1.23$ e Å⁻³
Extinction correction: SHELXTL
Extinction coefficient: 0.000081 (7)
Absolute structure: Flack (1983),
3084 Friedel pairs
Flack parameter: 0.699 (16)

Table 1
Selected geometric parameters (Å, °).

Eu1–S ⁱ	2.984 (3)	Eu5–S8	2.980 (2)
Eu1–S4	2.987 (3)	Eu5–S14	2.984 (3)
Eu1–S1	3.000 (3)	Eu5–S17 ^{vii}	3.027 (2)
Eu1–S14 ⁱⁱ	3.009 (3)	Eu5–S1 ^{viii}	3.039 (3)
Eu1–S10 ⁱⁱⁱ	3.028 (3)	Eu5–S10	3.073 (3)
Eu1–S2	3.056 (3)	Eu5–S9	3.082 (2)
Eu1–S13 ⁱⁱⁱ	3.081 (3)	Eu5–S15	3.110 (3)
Eu1···Eu2 ⁱ	4.0983 (12)	Eu5–S12	3.168 (3)
Eu1···Eu5 ⁱⁱ	4.1297 (11)	Eu5–S16 ^{vii}	3.208 (2)
Eu1···Eu2	4.1394 (12)	Eu5···Eu6	4.0621 (12)
Eu1···Eu4 ⁱ	4.1683 (11)	Eu5···Eu1 ^{viii}	4.1296 (11)
Eu2–S11 ^{iv}	2.944 (3)	Eu6–S1 ^{ix}	2.988 (3)
Eu2–S4	2.962 (3)	Eu6–S14	3.017 (3)
Eu2–S5	2.981 (2)	Eu6–S9	3.035 (2)
Eu2–S14 ^{iv}	2.982 (3)	Eu6–S8 ^{vi}	3.046 (2)
Eu2–S1	3.008 (3)	Eu6–S11	3.063 (3)
Eu2–S3	3.010 (3)	Eu6–S15 ^{vi}	3.095 (2)
Eu2–S12 ^v	3.209 (3)	Eu6–S17 ^{vii}	3.101 (3)
Eu2···Eu1 ^{vi}	4.0983 (12)	Eu6–S13	3.167 (3)
Eu2···Eu6 ^{iv}	4.1127 (11)	Eu6–S16 ^{vi}	3.197 (3)
Eu2···Eu3	4.1743 (11)	Eu6···Eu2 ^{ix}	4.1127 (11)
Eu3–S9 ⁱ	2.968 (2)	Eu6···Eu3 ^{vi}	4.1587 (11)
Eu3–S5	3.005 (3)	Sb1–S1	2.449 (2)
Eu3–S4	3.015 (3)	Sb1–S2	2.507 (3)
Eu3–S6	3.023 (3)	Sb1–S12 ⁱⁱ	2.664 (3)
Eu3–S8	3.024 (2)	Sb2–S3	2.426 (3)
Eu3–S17	3.051 (3)	Sb2–S4	2.464 (2)
Eu3–S15 ^v	3.132 (3)	Sb2–S2	2.598 (3)
Eu3–S16	3.235 (3)	Sb3–S6	2.431 (3)
Eu3···Eu4	4.0614 (12)	Sb3–S5	2.471 (2)
Eu3···Eu6 ⁱ	4.1587 (11)	Sb3–S7	2.587 (3)
Eu3···Eu4 ⁱ	4.1753 (12)	Sb4–S8	2.416 (2)
Eu4–S4 ^{vi}	3.005 (3)	Sb4–S7	2.537 (3)
Eu4–S5	3.009 (2)	Sb4–S10	2.644 (3)
Eu4–S8	3.010 (2)	Sb5–S11	2.436 (3)
Eu4–S9	3.044 (2)	Sb5–S9	2.440 (2)
Eu4–S15 ^{vii}	3.066 (3)	Sb5–S10	2.534 (3)
Eu4–S7	3.077 (3)	Sb6–S13	2.434 (3)
Eu4–S17	3.098 (3)	Sb6–S14	2.445 (2)
Eu4–S16 ^{viii}	3.157 (3)	Sb6–S12	2.520 (3)
Eu4–S13 ^v	3.382 (3)	S15–S16	2.091 (3)
Eu4···Eu5	4.1417 (11)	S16–S17	2.092 (3)
Eu4···Eu1 ^{vi}	4.1683 (11)		
S1–Sb1–S2	91.74 (9)	S8–Sb4–S10	90.92 (9)
S1–Sb1–S12 ⁱⁱ	91.51 (9)	S7–Sb4–S10	95.27 (9)
S2–Sb1–S12 ⁱⁱ	96.50 (9)	S11–Sb5–S9	94.14 (9)
S3–Sb2–S4	96.62 (9)	S11–Sb5–S10	103.87 (8)
S3–Sb2–S2	93.86 (8)	S9–Sb5–S10	90.28 (10)
S4–Sb2–S2	93.15 (9)	S13–Sb6–S14	95.92 (9)
S6–Sb3–S5	94.92 (10)	S13–Sb6–S12	100.90 (9)
S6–Sb3–S7	94.70 (7)	S14–Sb6–S12	94.00 (9)
S5–Sb3–S7	92.21 (9)	S15–S16–S17	113.34 (15)
S8–Sb4–S7	92.32 (9)		

Symmetry codes: (i) $x - 1, y, z$; (ii) $-x + \frac{1}{2}, -y - 1, z + \frac{1}{2}$; (iii) $x - \frac{3}{2}, -y - \frac{3}{2}, -z$; (iv) $-x + \frac{3}{2}, -y - 1, z + \frac{1}{2}$; (v) $x - \frac{1}{2}, -y - \frac{3}{2}, -z$; (vi) $x + 1, y, z$; (vii) $x + \frac{1}{2}, -y - \frac{3}{2}, -z$; (viii) $-x + \frac{1}{2}, -y - 1, z - \frac{1}{2}$; (ix) $-x + \frac{3}{2}, -y - 1, z - \frac{1}{2}$.

Inspection of the atomic positions reveals that they come in symmetry-related pairs (e.g. Eu1–Eu2, Eu3–Eu4 and Eu5–Eu6) and

the presence of additional pseudo-symmetry elements is confirmed by *PLATON* (Spek, 2003). The structure can be refined in space group *Pmnb* (alternative setting of the standard space group *Pnma*) to a residual of $R(F) = 0.11$ if a large number of systematic absence exceptions are excluded. The structure in *Pmnb* is identical to the structure in $P2_12_12_1$, except that the S_3^{2-} groups are in a disordered arrangement. We conclude, in agreement with Choi & Kanatzidis (2000), that the correct space group is indeed $P2_12_12_1$. The sole atom that breaks the mirror symmetry is S16. The crystal of $\text{Eu}_6\text{Sb}_6\text{S}_{17}$ which was examined proved to be a partial inversion twin. The highest peak and the deepest hole are located 0.98 and 1.77 Å from atoms Eu1 and S1, respectively.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT-Plus* (Bruker, 1999); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *XP* in *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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